

Synthesis and Antitumor Activity of a New Class of Pyrrazolo[4,3-e]pyrrolo[1,2-a][1,4]diazepinone Analogs of Pyrrolo[1,4]benzodiazepines (PBDs)

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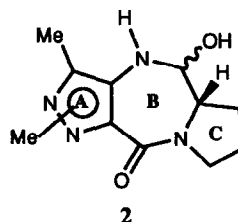
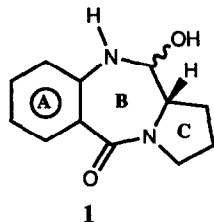
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Abstract: A new class of Pyrrolo[1,4]benzodiazepines (PBDs) analogs featuring a pyrazolo[4,3-e]pyrrolo[1,2-a][1,4]diazepinone ring system has been designed and synthesized. In these compounds the A-benzene ring, characteristic of PBDs, has been replaced by a dimethylpyrazole ring, a modification suggested by modelling studies performed on the PBD base structure. Biological evaluation revealed appreciable antitumor activity for compounds 14 and 15 (8.84-22.4 μ M) which encourages further investigation of the N⁶ or N⁷ alkyl pyrazole analogs.

Pyrrolo[1,4]benzodiazepines (PBDs) are potent antitumor antibiotics derived from various *Streptomyces* species showing interesting properties from both synthetic and biological standpoints¹. Natural products belonging to this family include Anthramycin, Tomaymycin, Neothramycins A and B, Sibiromycin and Chicamycin. The antitumor activity of these natural products is believed to involve the formation of a labile covalent amination linkage between the carbinolamine carbon (C11) of the antibiotics and the 2-amino group of guanine residues within the minor groove of DNA². Reaction with guanines in specific sequences results in DNA sequence specificity. The resulting DNA-antibiotic adduct inhibits DNA replication.



In the case of Anthramycin the precise structure of the drug-DNA adduct has been elucidated³, whereas for Tomaymycin NMR and fluorescence studies on the drug-DNA adduct⁴ have been made. Recent studies⁵ performed with dimeric PBD analogues have demonstrated that these new agents exhibit irreversible DNA cross-linking properties, and increased sequence specificity. The major limits to the clinical use of Anthramycin and other members of P[1,4]B family is their dose-limiting cardiotoxicity and tissue necrosis⁶. The P[1,4]B cardiotoxicity mechanism has demonstrated to be very similar to that of anthracyclines and it appears to be related to the formation of ortho quinone imine by oxidation⁷.

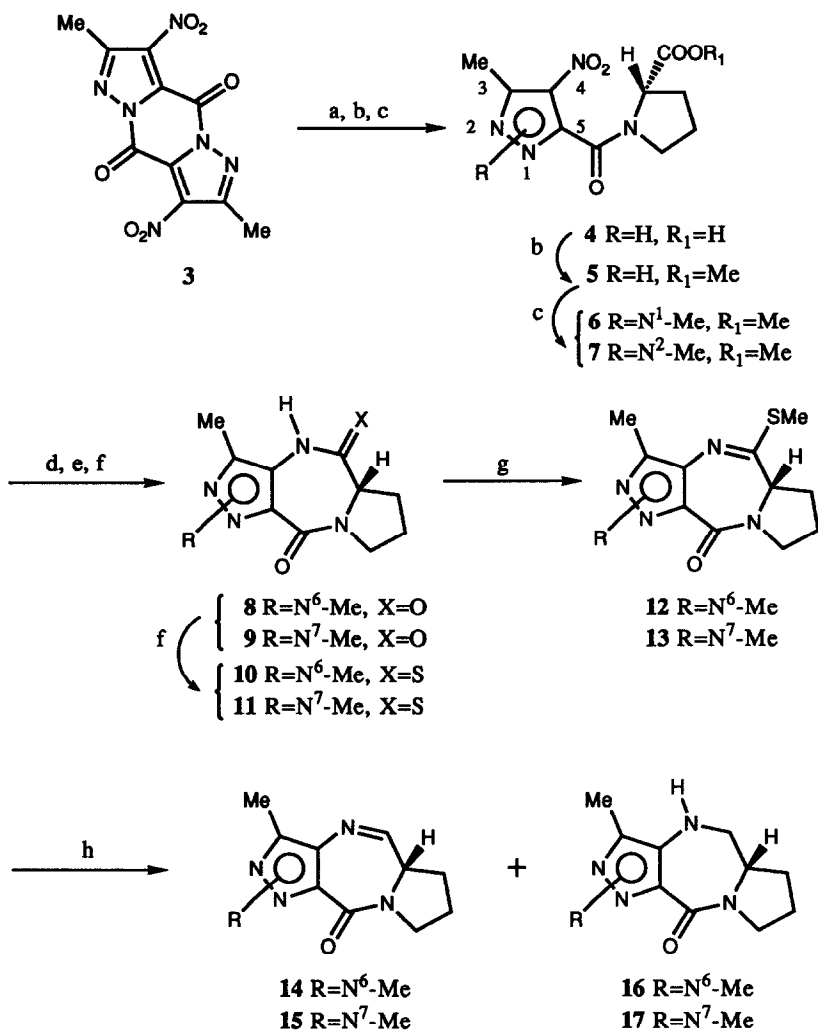
A rational approach to the development of clinically useful drugs in this series has been proposed, and some medicinal chemistry groups⁸ have begun to investigate synthetic methods for the preparation of rationally designed analogues to delineate more complete structure-activity relationships. In a project aimed to design new analogues of P[1,4]B lacking both cardiotoxicity and tissue necrosis, we have substituted the A benzene ring of the PBD skeleton (1) with a 1,3-dimethyl- or 1,5-dimethyl-pyrazole nucleus, a modification which, according to the CPK model proposed by Thurston and Hurley⁷, maintains all the structural requirements necessary for antitumor activity.

The unknown pyrazolo[4,3-e]pyrrolo[1,2-a][1,4]diazepinone ring system 2 could be obtained through standard synthetic routes established for PBD's involving direct coupling of pyrazolic diketopiperazine 3 with L-proline followed by reduction and cyclization to the dilactam and elaboration to the corresponding target compounds. Reaction of the readily available diketopiperazine 3 with L-proline in a 1:1 mixture of DMF-water in the presence of 1,1,3,3-tetramethylguanidine (TMG) furnished the acid 4 in good yield, which was in turn converted into the ester 5. Methylation of 5 with methyl iodide in the presence of sodium methoxide in methanol solution gave a 1:1 mixture of the corresponding 1,3-dimethyl and 1,5-dimethyl derivatives 6 and 7.

After chromatographic separation, the pure nitro derivatives 6 and 7 were both reduced with aqueous TiCl_3 ¹⁰ in methanol solution and cyclized *in situ* by sodium methoxide to the corresponding dilactams 8 and 9 in good overall yield. In order to introduce the imine functionality in the 1,4-diazepine ring of 8 and 9, we took advantage of the general and mild method developed by Kaneko *et al.*,¹¹ namely the aluminum-amalgam reduction of imino thioethers. Thiation of secondary amide function of 8 and 9 was achieved selectively with Lawesson's reagent to give the thioamides 10 and 11 which were in turn methylated to the corresponding imino thioethers 12 and 13 in fairly good yield. Treatment of 12 and 13 with Al-Hg in aqueous THF at 0-5°C for 24h gave a mixture of products which after purification led to the target compounds 14 (35%) and 15 (42%) along with minor amounts of the over reduction products 16 (15%) and 17 (12%). Compounds 14 and 15 were in the imine form, as was unambiguously established by ¹H and ¹³C NMR spectroscopy¹².

Antitumor activity of compounds 12-17 was evaluated *in vitro* on a L 1210 tumor cell line¹³. Compounds 14 ($\text{IC}_{50} = 22.4 \mu\text{M}$) and 15 ($\text{IC}_{50} = 8.84 \mu\text{M}$) were 100-1000 fold less potent than the natural product Tomaymycin ($\text{IC}_{50} = 0.01 \mu\text{M}$) and comparable to those of L-PAM ($\text{IC}_{50} = 11.6 \mu\text{M}$) used as a reference compound. Compounds 12,13 and 16,17, lacking the imine function characteristic of PBDs, were devoid of any antitumor activity. These results clearly indicate that the substitution of the A-ring of PBDs by a 1,3-dimethyl or a 1,5-dimethyl pyrazole ring still produces active compounds, whereas modification of the imine functionality results in loss of activity. This benzene/pyrazole substitution looks promising for the preparation of N⁶ and N⁷ alkyl derivatives to modulate the DNA binding in this new class of anthramycin analogs.

Scheme



a: L-proline, TMG; b: MeOH, H⁺; c: MeI, MeONa; d: TiCl₃; e: MeONa;
 f: Lawesson's reagent; g: MeI, K₂CO₃, THF; h: Al/Hg, THF/H₂O, 0°C.

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12. (10aS)-1,2,3,5,10a-hexahydro-6,8-dimethyl-pyrazolo[4,3-e]pyrrolo[1,2-a][1,4]diazepin-5-one
14: yellow oil; ^1H NMR (CDCl_3) δ : 1.94-2.02 (m, 2H); 2.26 (s, 3H); 2.28-2.35 (m, 2H); 3.51-3.58 (m, 2H); 3.6-3.7 (m, 1H); 4.07 (s, 3H); 7.34 (d, $J=3.8$ Hz, 1H); ^{13}C NMR: 10.6, 23.9, 29.9, 39.6, 45.9, 54.9, 127.8, 131.1, 144.1, 157.8, 158.9.
(10aS)-1,2,3,5,10a-hexahydro-7,8-dimethyl-pyrazolo[4,3-e]pyrrolo[1,2-a][1,4]diazepin-5-one
15: mp 139-140°C (EtOAc); ^1H NMR (CDCl_3) δ : 1.92-1.97 (m, 2H); 2.25 (s, 3H); 2.30 (m, 2H); 3.6-3.8 (m, 2H); 3.8-3.85 (m, 1H); 3.85 (s, 3H); 7.37 (d, $J=3.8$ Hz, 1H). 2D NOE (NOESY) experiments showed cross-peak between the N^7 -methyl ($\delta = 3.85$) and C^8 -methyl ($\delta = 2.25$) groups. ^{13}C NMR: 8.9, 24.1, 30.3, 37.3, 46.2, 55.3, 129.6, 136.1, 137.9, 159.9, 160.9.
13. Dr.G.Pezzoni and Dr.M.Bergomi, personal communication.